

was measured by Dr. Benjamin Rodríguez, Consejo Superior de Investigaciones Científicas Madrid, Spain.

REFERENCES

1. Melchior, H. (1964) *A. Englers-Syllabus der Pflanzenfamilien* 12th edn, Vol. 2, p. 172. Gebrüder Borntraeger, Berlin.
2. Delle Monache, G., Gonzalez Gonzalez, J., Delle Monache, F. and Marini-Bettolo, G. B. (1980) *Phytochemistry* **19**, 2025.
3. Locksley, H. D. and Murray, I. G. (1971) *J. Chem. Soc. C*, 1332.
4. Dreyer, D. L. (1974) *Phytochemistry* **13**, 2883.
5. Rama Rao, A. V., Venkatswamy, G. and Yemul, S. S. (1980) *Indian J. Chem.* **19B**, 627.
6. McCandlish, L. E., Hanson, J. C., Stout, G. H. (1976) *Acta Crystallogr. Sect. B* **32**, 1973.
7. Krishnamurthy, N., Lewis, Y. S. and Ravindranath, B. (1981) *Tetrahedron Letters* **22**, 793.
8. Hussain, R. A., Owegby, A. G., Parimoo, P. and Waterman, P. G. (1982) *Planta Med.* **44**, 78.

Phytochemistry, Vol. 22, No. 9, pp. 2090–2092, 1983.
Printed in Great Britain.

0031-9422/83 \$3.00 + 0.00
© 1983 Pergamon Press Ltd.

5-METHOXY-2,2,8-TRIMETHYL-10-SENECIOYL-2H,6H-BENZO(1,2-b; 5,4-b')DIPYRAN-6-ONE FROM *SPATHELIA WRIGHTII*

MARIA DIAZ, ALFRED PREISS*, HINRICH MEYER† and HELMUT RIPPERGER*

Institute of Botany, Academy of Sciences of Cuba, Havana, Cuba; *Institute of Plant Biochemistry, Academy of Sciences of the GDR, Halle (Saale), German Democratic Republic; †Department of Chemistry, Martin Luther University, Halle (Saale), German Democratic Republic

(Received 16 December 1982)

Key Word Index—*Spathelia wrightii*; Rutaceae; pyranochromone; 5-methoxy-2,2,8-trimethyl-10-senecieryl-2H,6H-benzo (1,2-b; 5,4-b')dipyrans-6-one.

Abstract—From leaflets of *Spathelia wrightii* a new pyranochromone has been isolated, the spectroscopic properties of which are in accordance with the structure 5-methoxy-2,2,8-trimethyl-10-senecieryl-2H,6H-benzo(1,2-b; 5,4-b')dipyrans-6-one (1).

From dried leaflets of *Spathelia wrightii* a pyranochromone was isolated in 2.9% yield. High resolution mass spectrometry revealed the elemental composition $C_{21}H_{22}O_5$. The structure was mainly deduced from NMR spectral information. The 1H NMR signals (Table 1) were assigned and partial structures recognized by comparison with chemical shift data and coupling constants of compounds containing the 2,2-dimethylchromene [1–3] and 2-methylchromone ring systems [4] or the senecieryl side chain [5]. In deuteriochloroform the signals of the C-14 and C-18 methyl groups coincide. Their membership to two spin systems was detected by decoupling experiments. The observed NOE values (Fig. 1) are only in agreement with one (1) of the 12 possible arrangements of the substituents at the benzene ring. In the NOE experiments, difficulties arose from the superimposition of the signals of H-14 and H-18 in deuteriochloroform and the neighbourhood of the signals of H-14 and H-19 in deuterobenzene, respectively. Irradiation on the signal at δ 2.18 in deuterobenzene showed that the C-18 methyl group does not produce a NOE. Therefore, the following measurements were carried out in deuteriochloroform considering that signal enhancement by irradiation on the signal at δ 2.24 has to be attributed exclusively to the C-14 methyl group. The ^{13}C NMR spectrum (Table 2), including the ^{13}C – 1H long-range coupling constants (Table 3), is

in agreement with structure 1. Signals were assigned by comparison with reported data for furochromones [6], mulberrochromene [7] and isobutenyl phenyl ketone [8] as well as by determination of the ^{13}C – 1H long-range coupling constants via selective decoupling experiments.

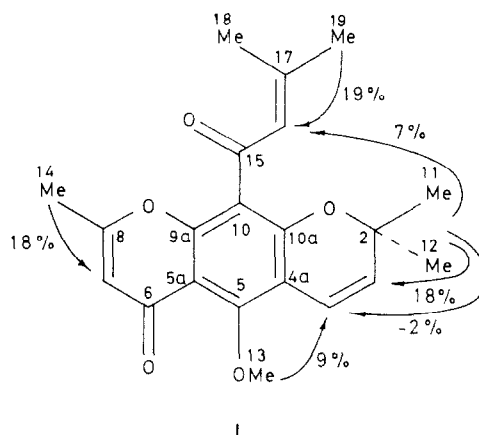


Fig. 1. NOE values for 5-methoxy-2,2,8-trimethyl-10-senecieryl-2H,6H-benzo(1,2-b; 5,4-b')dipyrans-6-one (1).

Table 1. ^1H NMR chemical shifts of compound 1*

Proton No.	CDCl_3	C_6D_6	Proton No.	CDCl_3	C_6D_6
3	5.69 (d)†	5.21	14	2.24 (d)	1.43
4	6.60 (d)	6.69	16	6.28 (qq)	6.25
7	5.96 (q)	5.74	18	2.24 (d)	2.18
11,12	1.46 (s)	1.18	19	1.97 (d)	1.53
13	3.89 (s)	3.90			

* δ -Values (ppm) measured from the solvent line and calculated relative to TMS ($\delta_{\text{TMS}} = \delta_{\text{CDCl}_3} + 7.24$ ppm, $\delta_{\text{TMS}} = \delta_{\text{C}_6\text{D}_6} + 7.15$ ppm).

† Multiplicities; $J_{3,4} = 10.2$ Hz, $J_{7,14} = 0.8$ Hz, $J_{16,18} = 1.3$ Hz, $J_{16,19} = 1.3$ Hz.

Table 2. ^{13}C NMR chemical shifts of compound 1*

Carbon No.	δ	Carbon No.	δ	Carbon No.	δ
2	78.0 (s)†	7	111.9 (d)	14	19.0 (q)
3	130.5 (d)	8	162.9 (s)	15	189.2 (s)
4	116.6 (d)	9a	155.9 (s)	16	127.0 (d)
4a	113.1 (s)	10	117.3 (s)	17	154.1 (s)
5	156.2 (s)	10a	154.6 (s)	18	20.8 (q)
5a	112.9 (s)	11,12	28.0 (q)	19	27.3 (q)
6	175.5 (s)	13	62.8 (q)		

* δ -Values (ppm) measured from the central solvent line (C_6D_6) and calculated relative to TMS ($\delta_{\text{TMS}} = \delta_{\text{C}_6\text{D}_6} + 128.0$ ppm).

† Multiplicities from the SFORD spectrum; under these conditions long-range couplings were not observed.

Table 3. ^{13}C - ^1H long-range coupling constants (Hz) for quaternary sp^2 -carbon atoms in compound 1

Carbon No.	Proton No.					
	3	4	7	13	14	16
4a (dd)*	8.5	1.8	—	—	—	—
5 (dq)	—	2.4	—	2.4	—	—
5a (d)	—	—	3.0	—	—	—
6 (d)	—	—	1.2	—	—	—
8 (dq)	—	—	6.7	—	6.7	—
9a (s)	—	—	—	—	—	—
10 (s)	—	—	—	—	—	not observed
10a (d)	—	4.3	—	—	—	—
15 (d)	—	—	—	—	—	3.6

* Multiplicities due to long-range couplings.

The 5-hydroxy compound corresponding to chromone (1), 10-senecioldipyranspatheliachromene, has been isolated from *Spathelia sorbifolia* and has a similar ^1H NMR spectrum [9].

EXPERIMENTAL

^1H and ^{13}C NMR spectra were recorded at 200.13 and 50.327 MHz, respectively. The NOE studies were performed on degassed samples of less than 5% concn (w/v) in CDCl_3 and C_6D_6 , respectively. Each resonance was integrated $\times 7$ in a homo inverse-gated decoupled and in a decoupled spectrum and the NOEs were calculated from the average areas. Accuracies of

NOE values are $ca \pm 1.5\%$.

Spathelia wrightii M. Vict. was collected in November at Yagrumajes River, Moa, Oriente, Cuba, and identified by Lic. Pedro Herrera. A voucher specimen is retained in the Herbarium of the Institute of Botany, Academy of Sciences of Cuba, Havana.

5-Methoxy-2,2,8-trimethyl-10-senecioldipyrans-6-one (1,2-b; 5,4-b')dipyrans-6-one (1). Dried and ground leaflets of *S. wrightii* were extracted with EtOH at room temp. Evaporation of the solvent *in vacuo* gave a residue which was partitioned between 0.5 N HCl and C_6H_6 -Et₂O (1:1). The residue of the organic phase was chromatographed over Si gel with Et₂O. From MeOH crystals with mp 126–127°. IR $\nu_{\text{max}}^{\text{neat}}$ cm^{-1} : 1660, 1634, 1611,

1582. UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm (log ϵ): 335 (3.87), 262 (4.71). EIMS, 70 eV m/z (rel. int.): 354.1455 (calc. for $\text{C}_{21}\text{H}_{22}\text{O}_5$: 354.1467; 55), 339.1189 (calc. for $\text{C}_{20}\text{H}_{19}\text{O}_5$: 339.1232; 100), 321.1164 (calc. for $\text{C}_{20}\text{H}_{17}\text{O}_4$: 321.1127; 6).

REFERENCES

1. Yamaguchi, K. (1970) *Spectral Data of Natural Products* Vol. 1, p. 65. Elsevier, London.
2. Gray, A. I., Waigh, R. D. and Waterman, P. G. (1975) *J. Chem. Soc. Perkin Trans. 1*, 488.
3. Tomimatsu, T., Hashimoto, M., Shingu, T. and Tori, K. (1972) *Tetrahedron* **28**, 2003.
4. González, A. G., Fraga, B. M. and Pino, O. (1974) *Phytochemistry* **13**, 2305.
5. Bohlmann, F., Zdero, C. and Ngo Le Van (1979) *Phytochemistry* **18**, 99.
6. Elgamal, M. H. A., Elewa, N. H., Elkhaisy, E. A. M. and Duddeck, H. (1979) *Phytochemistry* **18**, 139.
7. Wenkert, E. and Gottlieb, H. E. (1977) *Phytochemistry* **16**, 1811.
8. House, H. O. and Chia-Yeh Chu (1976) *J. Org. Chem.* **41**, 3083.
9. Taylor, D. R., Warner, J. M. and Wright, J. A. (1977) *J. Chem. Soc. Perkin Trans. 1*, 397.

Phytochemistry, Vol. 22, No. 9, pp. 2092–2093, 1983.
Printed in Great Britain.

0031-9422/83 \$3.00 + 0.00
© 1983 Pergamon Press Ltd.

5,7,3'-TRIHYROXY-6,8-DI-C-METHYL-4',5'-DIMETHOXYFLAVANONE FROM *ALLUAUDIOPSIS MARNIERIANA*

ZAFERA RABESA and BERNARD VOIRIN*

Département d'Ethnobotanique, Centre National de Recherches Pharmaceutiques, B.P. 702, Antananarivo, Madagascar; **Laboratoire de Phytochimie, Département de Biologie Végétale, 43 Boulevard du 11 novembre 1918, 69622 Villeurbanne, France

(Revised received 28 January 1983)

Key Word Index—*Alluaudiopsis marnieriana*; Didiereaceae; 5,7,3'-trihydroxy-6,8-di-C-methyl-4',5'-dimethoxy-flavanone; new natural compound.

Abstract—From bark and spines of *Alluaudiopsis marnieriana*, a novel flavanone has been isolated and identified as 5,7,3'-trihydroxy-6,8-di-C-methyl-4',5'-dimethoxyflavanone by UV, ^1H NMR and mass spectroscopy.

Previous studies of flavonoids of Didiereaceae have revealed different C- and O-methylated flavonols [1–7]. Of the four genera of this family, *Didierea*, *Decarya*, *Alluaudia* and *Alluaudiopsis*, only the latter lacks flavonols; instead it produces flavanones. We now report the structural elucidation of a new natural compound of this class, 5,7,3'-trihydroxy-6,8-di-C-methyl-4',5'-dimethoxyflavanone from *Alluaudiopsis marnieriana* Rauh, which was originally thought to be the 4'-hydroxy-3'-methoxy isomer [8].

The grey-violet fluorescence in UV light and the UV spectrum in methanol, band II main peak at 297 nm and band I at 348 nm, appeared typical of a flavanone [9]. This was confirmed by the presence of characteristic signals at δ 5.34 (1H, *dd*) and at 3.04 and 2.75 (1H, *dd*) corresponding to H-2 and H-3, respectively, in the ^1H NMR spectrum. Moreover, this spectrum exhibited signals at δ 1.97 (3H, *s*) and 1.99 (3H, *s*) and δ 3.71 (3H, *s*) and 3.80 (3H, *s*) typical for two C-Me and two methoxyls, respectively, and only two aromatic proton signals at δ 6.67 (1H, *d*, $J = 2$ Hz) and 6.69 (1H, *d*, $J = 2$ Hz) corresponding to two protons in the *m*-position. The positive results obtained after irradi-

ation of each aromatic signal indicated that these protons were located in the B ring at C-2' and C-6'. The mass spectrum exhibited a molecular ion peak at m/z 360 (85%) in accord with a flavanone containing three hydroxyl, two methoxyl and two C-methyl substituents ($\text{C}_{19}\text{H}_{20}\text{O}_7$, calc. 360.1209, found 360.1213). After fragmentation, a peak at m/z 207 ($[\text{M}]^+ - \text{B ring}$ according to ref. [10]) and high resolution measurement of peak B at m/z 180 ($\text{C}_{10}\text{H}_{12}\text{O}_3$ calc. 180.0786, found 180.0781) indicated that the B ring was substituted by one hydroxyl and two methoxyl groups. Of the two structural possibilities, 4'-hydroxy-3',5'-dimethoxy or 3'-hydroxy-4',5'-dimethoxy, the latter seemed more likely since in ^1H NMR spectrum the proton signals as well as those of methoxyl groups were separated. Nevertheless, this conclusion was confirmed by an NOE experiment; after irradiation of each methoxyl, only the irradiation on the methoxyl at δ 3.80 (C-5') gave rise to a significant result in the singlet at 6.69 (C-6'). In the A ring, two hydroxyls were located at C-5 and C-7, band II showing a bathochromic shift (43 nm) after addition of sodium methoxide in the UV spectrum [9]. The two remaining C-methyl groups were, thus, located at C-6 and C-8. High resolution measurements of ion fragments obtained after RDA reaction of peak D at m/z 181 ($\text{C}_9\text{H}_9\text{O}_4$ calc. 181.0501, found 181.0503) and

*To whom correspondence should be addressed.